

Answers Sheet

Speed and Density

1.

$$\text{Speed} = \frac{\text{distance}}{\text{Time}}$$

$$= 300 / 30$$

$$\text{Speed} = 10\text{m/s}$$

2.

$$= 140 / 4$$

$$\text{Pressure} = 35 \text{ N/m}^2$$

3.

$$\text{Density} = \frac{\text{mass}}{\text{volume}}$$

$$= 450 / 30$$

$$\text{Density} = 15 \text{ g/cm}^3$$

4.

$$\text{Volume} = \frac{\text{mass}}{\text{density}}$$

$$= 85 / 2.5$$

$$= 170 / 5$$

$$= 85\text{cm}^3$$

5.

$$\text{Speed} = \frac{\text{distance}}{\text{Time}}$$

$$\text{Speed} = 340 / 4.25$$

$$= 80 \text{ mph}$$

6.

$$\begin{aligned} \text{Area} &= \text{force} / \text{pressure} \\ &= 64 / 24 \\ &= \mathbf{2.67m^2} \end{aligned}$$

7.

$$\begin{aligned} \text{Mass} &= \text{density} \times \text{volume} \\ &= 1.5 \times 350 \\ &= \mathbf{525g} \end{aligned}$$

8.

$$\begin{aligned} \text{Time} &= \text{distance} / \text{speed} \\ &= 93 / 27 \\ &= \mathbf{3.4 \text{ hours}} \end{aligned}$$

9.

$$\begin{aligned} \text{Zohaib distance} &= \text{speed} \times \text{time} \\ &= 75 \times 2.3 \\ &= 172.5 \text{ miles} \\ \text{Shiraz Speed} &= \text{distance} / \text{time} \\ &= 172.5 / 2.583 \\ &= \mathbf{66.7 \text{ miles/hour}} \end{aligned}$$

b) If the distance was increased the speed would be higher.

(If the distance decreased the speed would be lower)

10.

First 267 miles

Time = distance / speed

Time = 267 / 80

= 3.33 hours

3 hours 20 mints (approx.)

So,

3 hours 20 mints + 5 hours 40 mints = 9 hours

Average speed = total distance / total time

= 400 / 9

= **44.4 mph**

11.

Speed = distance / time

= 2.1 / 12.583

= 0.1668... miles / min

Time = distance / speed

= 6 / 0.1668

= 35.97

35 mints 97 secs

b) It would take longer to run 6 miles
(The answer would be higher)

12.

Liquid A: Mass = density × volume

= 1.3 × 180

= 234g

Liquid C: Volume = mass / density

= 395 / 1.25

= 316cm³

Liquid B volume = Liquid C Volume – Liquid A Volume

= 316 – 180 = 136 cm³

Liquid B Mass = liquid C Mass – Liquid A Mass

= 395 – 234

= 161 g

Liquid B density = mass / volume

= 161 / 136

= **1.18 g / cm³**

13.

$$\begin{aligned}\text{Liquid A: Mass} &= \text{density} \times \text{volume} \\ &= 0.7 \times 200 \\ &= 140\text{g}\end{aligned}$$

$$\begin{aligned}\text{Liquid B: Mass} &= 1.1 \times 300 \\ &= 330\text{g}\end{aligned}$$

$$\begin{aligned}\text{Liquid C: Density} &= \text{total mass} / \text{total volume} \\ &= 140 + 330 / 200 + 300 \\ &= 470 / 500 \\ &= \mathbf{0.94 \text{ g / ml}}\end{aligned}$$
