insoluble Salts and precipitation

Solubility rules

Salt type	soluble
Sodium Potassium aluminium salts	always soluble
Nitrate	always soluble
chloride	soluble accept silver Chloride and lead chloride
sulphate	soluble except let barium and Calcium
carbonate and hydroxide	insoluble except Sodium Potassium aluminium

Making infallible salts precipitation

Making lead chloride insoluble

- Mix Lead Nitrate
- sodium chloride

Steps

1. Dissolve Lead Nitrate in water

2. dissolve sodium chloride in water
3. makes both lead chloride forms as solid
4. filter using filter funnel and paper
5. wash with the ionized water
6. dry in oven
7. lead chloride is insoluble and it will form the white precipitate